

previously:

- Clastic Sedimentary Rocks
- Chemical Sedimentary Rocks
 - Phosphates
 - Ironstones (Hematite, Limonite, Magnetite, Siderite, Goethite, Banded Iron Formations, and more)
 - Evaporites (Halite, Gypsum, Anhydrite, and many more)
 - Carbonates (Limestones, Dolostones, and more)
 - Cherts

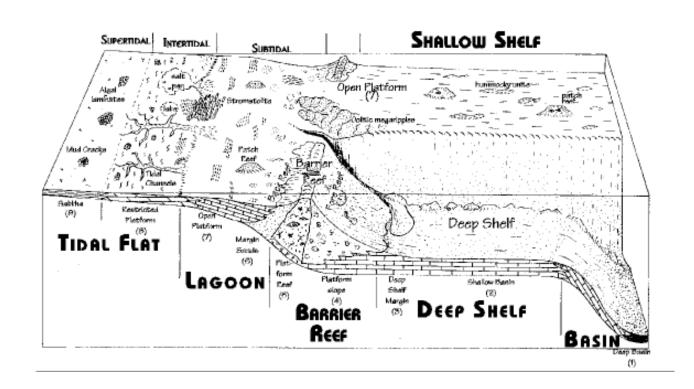
Carbonates

- Inorganic
 - form directly in water because of changing chemical equilibrium in solution
 - Travertine, Tufa, Oolitic Limestone
- Biochemical/Bioclastic
 - form because of the biological activity of an organism (biochemical) and successive mechanical weathering of remains (bioclastic)
 - Coral Reefs and Stromatolites, Fossiliferous Limestone, Coquina, Chalk, Micrite

Where do carbonates form?

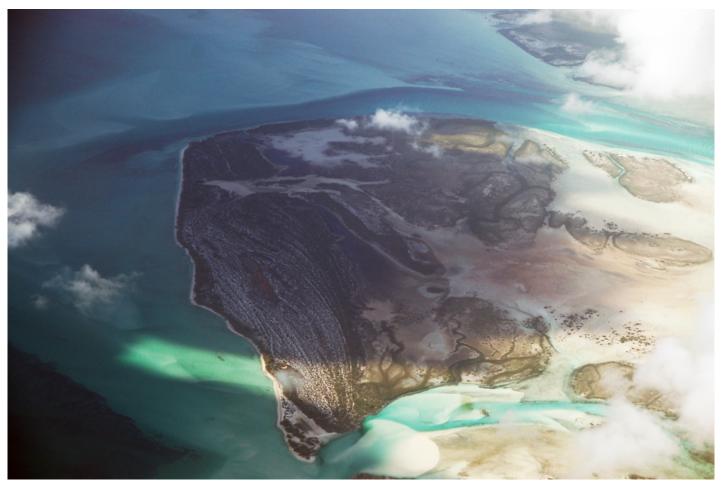
- While some form on land, most carbonates are deposited in the ocean
- On land:
 - Subaerial (travertine in caves, at waterfalls, at springs, cold but particularly hot springs)
 - lakes (tufa and travertine in alkaline and/or saline lakes)
 - carbonate sand dunes (along coastlines with carbonate sedimentation, such as Bahamas)
- Transitional environments:
 - Tidal flats (oolites, stromatolites)
 - Beaches (fossiliferous limestones, coquina)

Carbonate depositional environments



Bahamas: a modern carbonate shelf

micrites, oolites, carbonate sand dunes



Flying to Providence Island, Bahamas, from Miami, Florida © Alessandro Grippo

Where do carbonates form?

- In the shallow ocean (particularly if water is warm):
 - Lagoons, bays (fossiliferous limestones, oolites, micrite)
 - Open carbonate shelves in general (fossiliferous limestones, coquina, oolites, coral reefs, stromatolites)
 - Isolated platforms shallow water shelf sections separated from the main shelf (coral reefs, micrites, oolites)
- In the deep ocean:
 - Abyssal plain (chalk, micrite)

Cyclic ancient shelf carbonates



Salamayuca, Chihuahua, Mexico Mexico SR 45, Ciudad Juarez to Chihuahua, Chihuahua © Alessandro Grippo

Carbonates in shallow waters: neritic deposits

- neritic deposits are in general dominated by sediments that come form land (gravel, sand, silt, clay)
- in certain areas, particularly in shallow tropical waters, carbonate deposits are abundant
- We speak of a carbonate factory
 - Carbonate minerals containing CO₃
 - Marine carbonates primarily limestone CaCO₃
 - Most limestones contain fossil shells
 - Suggests biological origin
 - Ancient marine carbonates constitute 25% of all sedimentary rocks on Earth.

Coral Reefs

- Massive structure of carbonate
- Warm, crystal-clear, shallow waters
- Atolls, Barriers,Fringing Reefs

Stromatolites

- Fine layers of carbonate
- Warm, shallowocean, high salinity
- Cyanobacteria







What is a Coral Reef?

A reef is:

- a structure rising above the sea floor
- built by calcium carbonate secreting organisms
- that stand on the remains of their ancestors
- and are surrounded and often buried by the skeletal remains of the many small organisms that once lived on, beneath, and between them

Who builds a Coral Reef?

 A reef is almost entirely the physical expression of a community of organisms, growing in one place for an extended period of time

 These communities have changed dramatically over geologic time, so today's reefs can be very different from those of the past



The Great Barrier Reef of Australia

Modern Reefs

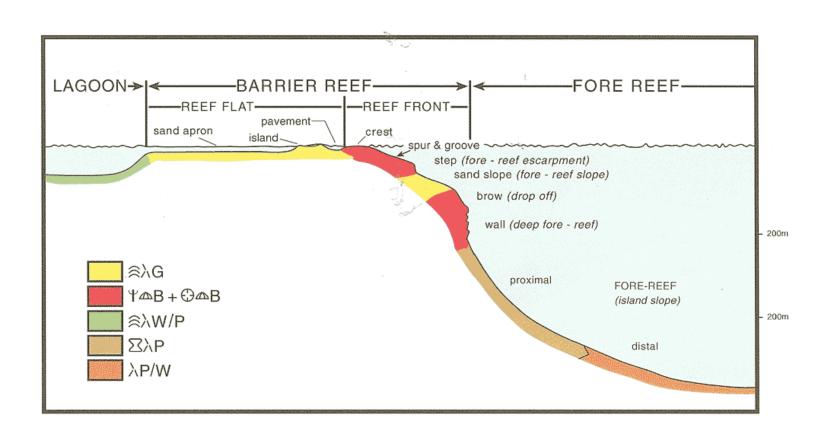
 While Coral Reefs are abundant today (Caribbean and Indo-Pacific), we must be aware of other in-place accumulations of skeletal calcium carbonate rising from the sea floor

 That includes reefs made of almost entirely of algae, banks of branching coral, and common skeletal sediment in deep water

Coral Reefs

- Today, the most widespread and massive reefs are constructed by corals and algae
- These organisms (either Cnidarian polyps or algae) require the following conditions for growth:
 - Tropical shallow waters
 - warmth favors secretion of carbonate
 - Shallow waters allow algae, either on their own or living in symbiosis with polyps, to make photosynthesis
 - Distance form sources of sediment
 - Rivers, coastlines, would be sources of fine sand and mud
 - Reef-building organisms are intolerant of fine sediment

Anatomy of a Coral Reef



What is a Stromatolite? from Cyanobacteria to Stromatolites

 Some filamentous cyanobacteria float as greenish scum on lake, streams, or ocean waters

 Others form "algal" mats on the seafloor that can trap sediment to produce distinctive 3-D structures (stromatolites)



Modern Stromatolites from Shark Bay, Australia

Stromatolites

in four "simple" steps

- Accretionary organosedimentary structures the structure build up (accretes), and forms a structure through interaction of biological and physical processes
- 2 commonly thin-layered, megascopic, and calcareous made of thin, stacked laminae, visible to the naked eye, partially composed of calcium carbonate minerals
- 3 produced by the activity of mat-building communities of mucilagesecreting microorganisms
 - microscopic organisms living together generate mats, or layers by secreting sticky gelatin-like slime
- mainly filamentous photoautotrophic prokaryotes such as cyanobacteria

most organisms are developing threads (and not spheres), are photosynthetic, are Bacteria and Archaea, and most of the times are cyanobacteria

Fossil Stromatolites

from Glacier National Park, Montana







Fossiliferous Limestones

- Any type of limestone, made mostly of CaCO₃ (as calcite or aragonite) that contains an abundance of fossils or fossil traces
- The fossils in these rocks may be of macroscopic or microscopic size

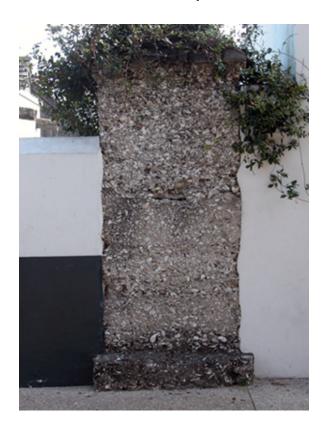


Cretaceous fossiliferous limestone with recrystallized belemnites
Panther Junction, Big Bend National Park, Texas

© Alessandro Grippo

Coquina

- A type of limestone formed almost entirely of sorted and cemented fossil debris, most commonly coarse shells and shell fragments
- Indicates deposition in a high-energy environment







Coquina was used to build the fortified city of San Agustin in Spanish Florida. See the whole story at

https://www.nps.gov/casa/learn/historvculture/coguina-the-rock-that-saved-st-augustine.htm

Saint Augustine, Florida © Alessandro Grippo

More carbonates?

Yes, what happens in deep waters?

 High pressure, low temperatures, there should not be any carbonate!

To be continued...