Introduction to **EARTH HISTORY**

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Earth History: does it matter?

- Desire to know about our world and to understand where we come from
- Necessity to find resources for our society
- Lessons from the past
 - Past changes can help foresee the future

Geology's contribution to human thought

Deep time

- Geology shows that Earth is 4.6 billion years old
- Humans have been around for an incredibly smaller amount of time

"Man has been here 32,000 years. That it took a hundred million years to prepare the world for him is proof that is what it was done for. I suppose it is, I dunno. If the Eiffel Tower were now representing the world's age, the skin of paint on the pinnacle-knob at its summit would represent man's share of that age; and anybody would perceive that that skin was what the tower was built for. I reckon they would, I dunno" (Mark Twain)

Resources: water, oil, metallic ore, etc.



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The Rammelsberg mines complex in Goslar, Germany

Zinc, iron, copper, silver, gold and more have been extracted for centuries at this location at the foot of the Harz Mountains in northern Germany

Lessons from the past



2008



2015

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Landslide in shales and gypsum, Borgo Tossignano, Bologna, Italy

EARTH HISTORY: arrow or cycle?

Arrow

 History is an irreversible sequence of unrepeatable events. Time has a direction

Cycle

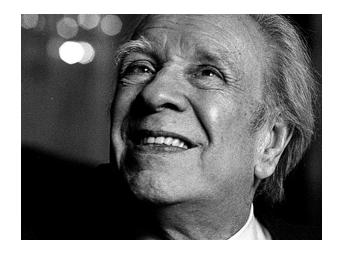
 Apparent motions are part of repeating cycles, and differences of the past will be realities of the future. Time has no direction

(Stephen Jay Gould, 1987)

"If space is infinite, we may be at any point in space.

If time is infinite, we may be at any point in time"

Thomas Burnet



Jorge Luis Borges's paradox in "The Book of Sand"

A beginning and an end? It is an arrow

Repetitions? It is a cycle

What if the cycles are slightly different, so that every repetition is actually not the same? In this way we know where we are in space, we know where we are in time.

EARTH'S SYSTEM TODAY

- A rocky physical body
 - Core, Mantle, Oceanic Crust, Continental Crust
- The Water
 - Hydrosphere
 - Salt waters: Oceans
 - Fresh waters: Glaciers, Groundwater, Rivers, Lakes
 - Water vapor in the atmosphere
- The Air
 - Atmosphere
 - 79% N₂, 21% O₂, traces of CH₄, CO₂, NO₂, SO₂, H₂O and other gases
- Living organisms
 - Prokaryotes (Archaea, Bacteria)
 - Eukaryotes (Protists, Fungi, Plants, Animals)

EARTH'S SYSTEM, always like today?

ROCKS

- formed from molten planet
- rocks abundance, type, distribution changed over time

WATER

- not present at beginning
- oceans became saltier in time, up to today's 34.7 ppm (3.47%)
- glaciers were more or less abundant than today at different times

AIR

- more greenhouse gases, no O₂ at beginning
- amount of O₂ variable in the past

- no organisms found on any other planet
- all organisms interfere with Earth
- organisms change in time (evolution)

Where do the components come from?

- Solid ROCKS formed when Earth cooled
- Liquids and gases were originally released during volcanic eruptions (WATER and ATMOSPHERIC GASES)
 - Water and air were retained thanks to gravitational force
- LIFE started as soon as Earth's surface was solid

In Oceans

 salt water; salinity is variable but the types and ratios of salt are identical all over the world in open oceans; mostly Na⁺ and Cl⁻ ions



The Pacific Ocean: Howe Sound, by Vancouver, B.C. Canada

- In Rivers
 - fresh water; composition may vary, depending on source



The Virgin River in Zion National Park Springdale, Utah

• In Lakes

fresh water or salt water; if salty, type of salt may

vary



An ephemeral, saline lake: Badwaters Death Valley National Park, California

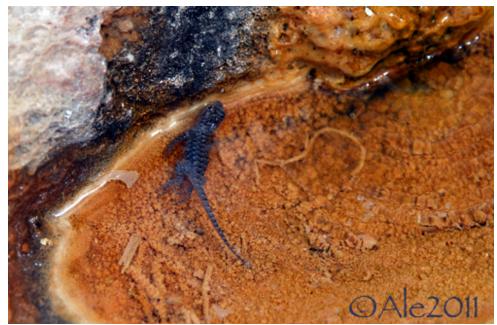


Convict Lake, Sierra Nevada, California



Mono Lake, Lee Vining, California

- As Groundwater
 - fresh water; can be salty close to ocean or at depth
 - temperature may vary: geothermal waters



Hot groundwater upwells at this location, killing animals who fall into pools, and causing deposition of calcium carbonate (orange crystals)

Hot Springs by Bridgewater, California

- As Ice
 - Ice on continents
 - Ice pack (sea ice), can contain up to 0.8% salt
 - Alpine ice



Glacier Bay, Gustavus, Alaska

- As Water Vapor
 - Atmospheric moisture, usually "distilled" water
 - Percentage can vary

- In Rocks
 - directly as water (e.g. Gypsum, CaSO₄*2H₂0)
 - trapped in clay minerals
 - as OH⁻ ions that can promptly react with H⁺ ions
- In Living Organisms
 - Plants (evaporation and transpiration processes)
 - Animals

AIR

• Nitrogen: 78%

• Oxygen: 21%

other gases: ~ 1%

- (CO₂, NO₂, CO, NO, CH4, H₂O, and others)

in orange: greenhouse gases

AIR

- Greenhouse gases:
 - Solar radiation can penetrate greenhouse gases
 - Radiation is reflected back by Earth, but at a different frequency
 - Greenhouse gases trap the radiation emitted by Earth
 - More greenhouse gases in the atmosphere imply warmer temperatures

- Unique to Earth
- Appeared as soon Earth developed a solid crust
- Started as:
 - chemosynthetic
 - prokaryotic
 - mitotic
 - single celled

- Evolved in time into:
 - photosynthetic
 - eukaryotic
 - meiotic (sexual reproduction)
 - colonial
 - from colonial to multicellular
 - from simple (e.g. sponges) to specialized (e.g. plants and animals)

- Organisms modify the environment
 - Photosynthesis
 - added oxygen to atmosphere
 - oxygen provides more energy as a fuel
 - life development sped up
 - Carbonate reefs
 - today they are limited, more developed in the past
 - Human activity

Photosyntesis

solar energy

$$6CO_2$$
 + $6H_2O$ \rightarrow $C_6H_{12}O_6$ + $6O_2$
carbon dioxide + water \rightarrow sugar + oxygen

Respiration

$$C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$$

sugar + oxygen \rightarrow carbon dioxide + water

release of (solar) energy stored in sugar

- Destruction vs. Preservation of Organic Matter
 - oxygen is needed to burn sugars
 - upon death, bodies are recycled by scavengers, decomposers, who all need oxygen
 - if there is no oxygen, organic matter (the sugars) do not decompose and are preserved, albeit modified
 - this preserved organic matter that escaped decomposition makes for hydrocarbons (oil and natural gas)